

CHUKA



UNIVERSITY

UNIVERSITY EXAMINATIONS

EXAMINATION FOR THE AWARD OF DEGREE OF BACHELOR
EDUCATION SCIENCE AND BACHELOR OF SCIENCE

PHYS 315: THERMAL AND STATISTICAL PHYSICS

STREAMS: BED (SCI) and B.Sc

TIME: 2 HOURS

DAY/DATE: THURSDAY 7/12/2017

11.30 A.M - 1.30 P.M.

INSTRUCTIONS:

- Answer Question One in Section A and any other Two Questions in Section B
- Do not write anything on the question paper
- This is a closed book exam, No reference materials are allowed in the examination room
- There will be No use of mobile phones or any other unauthorized materials
- Write your answers legibly and use your time wisely

SECTION A

QUESTION ONE

- a. Distinguish the following terms as used in thermodynamics [3 Marks]
- Microcanonical ensemble
 - Canonical ensemble
 - Grand canonical ensemble
- b. Give 2 statement of 3rd law of thermodynamics. [4 Marks]
- c. The macrostate that is most stable contains the majority microsities. Explain this statement.
- d. Find the number of ways in which two particles can be distributed in six states is,
- The particles are distinguishable [2 Marks]
 - The particles are indistinguishable and obey Bose-Einstein statistics [2 Marks]
 - The particles are indistinguishable and only one particle can occupy one state [2 Marks]
- e. Show that for every large number, the sterling's approximation is $n! = n \ln n - n$ given $n! = n(n - 1)(n - 2) \dots$ [3 Marks]
- f. Differentiate between Bosons and Fermions [4 Marks]

PHYS 315

- g. State the principle of equipartition of energy theorem [1 Mark]
- h. State 3 postulates of Maxwell-Boltzmann distribution [3 Marks]
- i. The equilibrium state is the highest entropy of a system, explain this statement [2 Marks]
- j. Given that ψ is antisymmetric such that $\psi_a = \sum_p (-1)^p P [\psi (1,2,3 \dots N)]$ where p is the permutatic operator, write the linear combination for 3 particles.

QUESTION TWO

- a. Taking S and V to be Independent variables with $x=S$ and $y=V$, derive the Maxwell's thermodynamic relation $\left(\frac{\partial T}{\partial V}\right)_p = -\left(\frac{\partial P}{\partial S}\right)_T$ stating from the relation $dU = Tds - pdV$ for an infinitesimal reversible process. [13 Marks]
- b. Using the thermodynamic relation $\left(\frac{\partial S}{\partial V}\right)_T = -\left(\frac{\partial P}{\partial T}\right)_V$ derive the Stefan Boltzmann law of radiation. [7 Marks]

QUESTION THREE

- a. Given that quantized energy is $E_j = \frac{\hbar^2 \pi^2}{2mL^2} (n_x^2 + n_y^2 + n_z^2)$ and that the partition function z is given by $Z = \sum_i e^{-\frac{E_i}{kT}}$ show that the partition function for Maxwell-Boltzmann distribution can be expressed as $Z = V \left(\frac{2m\pi kT}{\hbar^2}\right)^{\frac{3}{2}}$ where $V = L^3$. [10 Marks]
- b. Consider an ideal gas that contains N molecules with continuous distribution of molecular energies in which the Maxwell's distribution law is given by, $n(E)dE = g(E)e^{-\alpha} e^{-\beta} dE$ where, $\beta = \frac{E}{kT}$ show that the energy level of an ideal gas molecule is $E = \frac{3}{2} kT$ [10 Marks]

QUESTION FOUR

- a. If n is the number of conduction electron per unit volume and m is the electron mass, show that the Fermi energy is given by $E_f = \frac{\hbar^2}{8m} \left(\frac{3n}{\pi}\right)^{\frac{2}{3}}$ [9 Marks]
- b. Suppose n_1 particles occupy the first energy level with energy E_1 , n_2 occupy the second energy level with energy E_2 . Show that the number of ways n_i particles can be distributed into g_i cells in Fermi-Dirac distribution is given by, $n_j = \frac{1}{e^{-\alpha} e^{\beta \epsilon_j} + 1}$ [11 Marks]

PHYS 315

QUESTION FIVE

a. What are the differences between variable as used in microscopic and macroscopic descriptions. [4 Marks]

b. Given that $dU = Tds - pdV$ derive the Maxwell's equation $\left(\frac{\partial S}{\partial p}\right)_T = -\left(\frac{\partial v}{\partial T}\right)_p$ (9 Marks)

c. For n_i particles and g_i states show that the Bose Einstein distribution law is given by,

$$n_j = \frac{1}{e^{-\alpha} e^{\beta \epsilon_j} - 1} \quad [7 Marks]$$

.....